

# **Design and Analysis of a New Multigeneration System for a** Hospital

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I. Introduction As many developing countries strive toward industrialization and economic growth, they often use fossil fuels to facilitate their growth and sustain their current society	II. Analysis The electrolysis chemical reactions used to model hydrogen and oxygen production are presented as follows:	
<ul> <li>Jamaica primarily consumes fossil fuels.</li> </ul>	Anode: $H_2 O \rightarrow 2H^+ + \frac{1}{2}O_2 + 2e^{-1}$ (1)	
Other (Wind, Solar, Hydro, etc.), 1.9% Biofuels and waste, 7.2%	Cathode: $2H^+ + 2e^{-1} \rightarrow H_2$ (2)	
Natural Gas, 23.7%	Overall: $2H_2 O \to H_2 + \frac{1}{2}O_2$ (3)	



Jamaica has tremendous solar potential, due to its tropical climate.

Jamaica also has the potential to produce electricity and other useful outputs from sugar cane crops. Several studies suggest sugarcane bagasse in Jamaica can be used for electricity production within a cogeneration system [2]. Flue is used to produce



- Parishes such as Trelawny would benefit greatly from electricity production through bagasse, given that most of Jamaica's sugar cane crops are grown here, and power outages occur more frequently within this region.
- Power outages within Trelawny pose a serious safety risk to hospital patients.
- The proposed multigeneration system will be designed to produce hydrogen, and other useful outputs such as oxygen for patients, natural gas, hot water, electrical power, and space cooling.



The Sabatier chemical reaction between the captured carbon dioxide and a portion of the generated hydrogen is presented as follows:

$$CO_2 + 4H_2 \rightarrow CH_4 + H_2O \ \Delta H = -165 \frac{kJ}{mol}$$
 (4)

### **III. Results and Discussions**

The overall performance of the system is largely dictated by the performance of the PVT sub-system. The amount of electrical power generated by the PVT sub-system will dictate the amount of hydrogen, oxygen, and methane produced.



Peak Daily Solar Power Generated **-0**-02 

Figure 4. Comparison between H<sub>2</sub>, CH<sub>4</sub>, and O<sub>2</sub> production rates monthly

All of the useful outputs generated by the system are summarized in Table 1.

#### Table 1. System Performance

Useful Output	Value
Average Peak H <sub>2</sub> Production Rate	1.51e-3 kg/s
Average Peak CH <sub>4</sub> Production Rate	6.02e-3 kg/s

Average Peak O <sub>2</sub> Production Rate	2.41e-2 kg/s
Hot Water Production Rate	9.23e-3 kg/s
Net Power Output	367 kW
Cooling Load	73.25 kW

### **IV. Conclusions**

The system produces H<sub>2</sub>, O<sub>2</sub>, and CH<sub>4</sub> through PEM electrolysis and the Sabatier chemical reaction. The average peak oxygen production rate of 2.41e-2 kg/s has the benefit of helping patients who are connected to ventilators. Additionally, the system can produce a net hydrogen production rate of 1.51e-3 kg/s, and satisfy many other needs for the hospital such as space cooling, hot water, electricity, and natural gas/ methane.

#### References

[1] International Energy Agency, "Energy system of Jamaica." [Online]. Available: https://www.iea.org/countries/jamaica

[2] R. Contreras-Lisperguer, E. Batuecas, C. Mayo, R. Díaz, F. J. Pérez, and C. Springer, "Sustainability assessment of electricity cogeneration from sugarcane bagasse in Jamaica," J. Clean. Prod., vol. 200, no. 2018, pp. 390–401, 2018, doi: 10.1016/j.jclepro.2018.07.322.